SOV/126-7-3-41/44

On the Causes of Brittleness of Certain Copper Alloys

ASSOCIATION: Ural'skiy gosudarstvennyy universitet imeni A.M. Gor'kogo (Ural State University imeni A.M. Gor'kiy)

SUBMITTED: December 30, 1958

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Card 4/4

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

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18.1100 6689 sov/126-8-1-8/25 18.9200 AUTHORS: Arkharov, V. I. and Kralina, A.A. On the Influence of Palladium Additions to Iron on its TITLE: Hydrogen Permeability, PERIODICAL: Fizika metallov i metallovedeniye, 1959, Vol 8, Nr 1, pp 45-52 (USSR) ABSTRACT: From a consideration of a number of papers (Refs,1-6) it can be assumed that intercrystalline diffusion hydrogen will proceed to a greater extent in iron containing palladium than in iron without palladium. A similar influence of palladium is possible also in certain iron-base alloys. The authors have investigated diffusion of hydrogen through polycrystalline iron-base alloys containing palladium and compared it with alloys of the same composition but without palladium at identical large and small grain sizes. The following materials were used for experiments: 1) Armco iron, 2) iron containing 0.5% palladium, 3) austenitic iron-chrome-nickel (18% Cr, 7.5% Ni) and 4) an austenitic alloy with the same chromium and nickel content as in 3 but with an addition of 0.5% palladium. The basis of all alloys was Armco Card 1/6 iron. After melting and casting, specimens of the above

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On the Influence of Palladium Additions to Iron on its Hydrogen

alloys were forged into rods of 10 \times 10 mm cross-section Permeability which were subsequently rolled into plates, 0.25 mm thick (in two stages with inter-annealing at 600°C for 10 hours). The difference in thickness of the plates did not exceed 0.01 mm. Specimens, 12 x 12 mm, were cut out from the plates and were subjected to preliminary annealing at 600°C for 50 hours. Subsequently annealing was carried out in order to obtain fine or coarse grain size. specimens were annealed in vacuum for 10 hours at various temperatures which were so chosen as to obtain series of specimens of all four alloys with identical grain size; in one series a "fine" grain size $(20-25 \mu)$ and in another series a "coarse" grain size (280-290 µ) was obtained. In the austenitic alloys specimens were also obtained with "particularly coarse" grain size (560-580 µ). The annealing temperatures are indicated in Table 1. The condition of the surfaces of the specimens was characterized by the fact that on one side a metallographic section was prepared, whereas the other was allowed to remain in the same condition as after cold rolling and vacuum annealing.

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On the Influence of Palladium Additions to Iron on its Hydrogen

Experiments for the hydrogen permeability of the Permeability specimens were carried out by a special method. A short glass tube was luted to the plate specimen under investigation; thus a small container for an electrolytic bath was obtained. A 1% aqueous solution of sulphuric acid was used as the electrolyte. The specimen forming the bottom of the container was the cathode and a platinum plate (3 x 8 mm) the anode. The latter was placed vertically in the centre of the container, the lower end of the anode being 2 mm above the bottom. During electrolysis the anode was rotated around its vertical axis. Electrolysis was carried out at a current density of 0.22 amps/dm². The lower side of the specimen The lower side of the specimen closely adhered to the orifice of a gas analyser. The diameter of this orifice (10 mm) determined the surface area of the specimen (0.785 cm²) in direct contact with In this portion of its surface the specimen acted as a diffusion membrane for hydrogen, the gas analyser. forming at the cathodes during electrolysis; this Card 3/6 diffuses across the specimen into the inner part of the

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sov/126-8-1-8/25

On the Influence of Palladium Additions to Iron on its Hydrogen Permeability

gas analyser. The apparatus is shown in Fig 1. It has two chambers, one of which communicates with the space into which the hydrogen gets after diffusing through the specimen; the other is a closed chamber filled with air. In each chamber a calibrated platinum wire spiral of 0.02 mm diameter is placed. These spirals are connected to an electric circuit by a bridge system and they are heated by the current passed through both parallel arms of the system. A zero reading of the galvanometer in series with the bridge corresponds to identical atmosphere composition (air) in both chambers. As soon as some quantity of hydrogen appears in the first chamber the heat emission by the platinum spiral in this chamber is intensified, the temperature of the spiral, and hence its resistance, change, and the galvanometer shows a deflection. The authors used the galvanometer M-21 having a sensitivity of 10-9 A. The apparatus was first calibrated according to hydrogen-air mixtures of known composition. experiments in which the hydrogen permeability of the alloys investigated was determined, the galvanometer

Card 4/6

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66891 sov/126-8-1-8/25

On the Influence of Palladium Additions to Iron on its Hydrogen Permeability

readings were registered at even time intervals and a graph for the increase in hydrogen content in the catharometer (i.e. the quantity of hydrogen diffused through the specimen) with electrolysis time was plotted with the help of the graduation curve. The experimental results are shown in Figs 2-6 in the form of graphs showing the dependence of the quantity of hydrogen which had diffused through the specimens on the time of electrolysis. The measured values of the tangent of the angle of inclination of the straight portions of graphs in Figs 2-6 are shown in Table 2. The authors arrived at the following conclusions:

- 1) The hydrogen permeability of iron alloys increases
- when 0.5% palladium is added.
- 2) The increase in hydrogen permeability caused by the addition of palladium is due mainly to the intercrystalline
- 3) The hydrogen permeability in the mass of crystals seems to be due, to a considerable measure, to inter-block joints in the sub-crystal structure.

Card 5/6

On the Influence of Palladium Additions to Iron on its Hydrogen Permeability

2. 在大型工程的设置,1984年的1985

4) The influence of palladium additions on the hydrogen permeability of iron alloys is due to the internal adsorption of palladium.

There are 6 figures, 2 tables and 7 references, 6 of which are Soviet and 1 English.

ASSOCIATION: Institut fiziki metallov AN SSSR (Institute of Physics of Metals, Ac.Sc., USSR)

4

SUBMITTED: March 18, 1959

Card 6/6

"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8

THE THE PROPERTY OF THE PROPER

AUTHOR: Arkharov, V.I. 50V/126-8-2-7/26

TITLE: Kinetics of the Reaction Diffusion in Systems With Several

Intermediate Phases. I. Macroscopic Calculation for a

Binary System

SOURCE STREET, STREET,

PERIODICAL: Fizika metallov i mctallovedeniye, 1959, Vol 8, Nr 2,

pp 193 - 204 (USSR)

ABSTRACT: The kinotics of reactions involving diffusion are

complicated by, among other factors, the fact that one of the components increases partly on account of the adjacent layer. The author first considers the general picture of diffusion in a scaling reaction, where a series

of oxides can be formed. He then discusses initial conditions for calculating the rate of growth of scale layers at their boundaries for five possible cases,

depending on the relation between the ratio of the amount of metal retained by chemical reaction at the boundary to the quantity of oxygen retained and the ratio of the

quantities of components there. This leads to solutions for a multi-layer scale, with values of coefficients

Card1/2 capable of determination by observation of scale-layer

SOV/126-8-2-7/26

Kinetics of the Reaction Diffusion in Systems With Several Intermediate Phases. I. Macroscopic Calculation for a Binary System

growth. By more refined experimentation, making use of macroscopic particles of material not involved in reaction diffusion which is placed in some form on the specimen surface, the metal and oxygen diffusion coefficients can be determined separately from the kinetic curves. Normally, reaction diffusion kinetics determined from gain in weight of the specimen and its loss in weight when the scale is removed. The author shows how the "effective" reaction - diffusion coefficients can be found, including the particular case of a double-layer scale.

There are 3 references, 2 of which are Soviet and 1 German.

ASSOCIATION: Institut fiziki metallov AN SSSR (Institute of Physics of Metals of the Ac.Sc.USSR)

SUBMITTED: April 27, 1959

Card 2/2

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

18.7530 AUTHORS: 66236 Arkharov, V.I. and Blankova, Ye.B. sov/126-8-3-20/33 TITLE: Investigation of Reaction Diffusion in Binary "Metal-Gas" PERIODICAL: Fizika metallov i metallovedeniye, 1959, Vol 8, Nr 3, ABSTRACT: The mechanism of reaction diffusion depends mainly on the type of chemical combination between the reacting elements, structural characteristics of the phases formed and the size ratio of the defluxing particles. The authors suggest that for studying the diffusion mechanism in binary systems the reactions of elements within one period with a single second component, which is then successively changed on the same principle, should be investigated. They give results obtained for binary systems formed by transition elements of the IV period with sulphur. Diffusion annealing was effected at 400 to 1000°C in ampoules evacuated to 10-3 - 10-4 mm Hg, designed in such a way that the specimen was in contact

Card 1/2

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with sulphur vapour (1 atm approximately) in one part, the sulphur being in another part of the same ampoule.

Macro and micro-structures of the scale, its phase

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Investigation of Reaction Diffusion in Binary "Metal-Gas" Type Systems. I

composition, absence or presence of texture were studied and experiments with inert markers of platinum were also carried out. The results for Ti, V, Cr, Mn, Fe, Co and Ni with S are tabulated. The authors conclude that in all these systems the diffusion mechanism is broadly similar, both metal and sulphur diffusing. The diffusion rates of metal and sulphur increase to roughly the same extent with rising temperature. The rate of scale growth under given conditions increases from Ti-S to NiS that of Co-S, however, being anomalously low. There are 1 table and 16 references, 2 of which are Soviet and 14 Western.

ASSOCIATION: Ural'skiy gosudarstvennyy universitet im A.M. Gor'kogo (Urals State University im A.M. Gor'kiy)

SUBMITTED: March 31, 1959

Card 2/2

4

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

SOV/126-8-4-10/22 Arkharov, V.I., and Blankova, Ye.B. Investigation of Reaction Diffusion in Binary Systems 18.7530 AUTHORS: metallovedeniye, 1959, Vol 8, Nr 4, of the "Metal-Gas" Type. TITLE: PERIODICAL: Fizika metallov i ABSTRACT: The authors have given (Ref 1) the main facts associated The authors have given (ner 1) one manual formed by with the diffusion mechanism in systems formed by with the diffusion mechanism in systems formed by with the diffusion alamants with sulphur. fourth-group transition elements with sulphur. follow this with the investigation (based on their own IOLIOW this with the investigation (based on their own and published work) of systems of the same metals but with selenium or tellurium. The results on the mechanism of reaction-diffusion in the systems Ti-Se, V-Se, Cr-Se, Mn-Se, Fe-Se, Co-Se, Ni-Se, are shown in Table 1; temperature ranges (two for each system), number of temperature ranges (two for each system), number of temperature characteristics of the diffusion mechanism. macroscopic characteristics of the diffusion mechanism, corresponding information for the systems with Te.

In the Cr-Se, Mn-Se, Fe-Se and Ni-Se systems (as in the nature of texture in the scale layer. corresponding systems with sulphur) a diffusion of the metal and selenium occurs over the whole temperature range (up to 1000 oC); but with Ti-Se, V-Se and Co-Se Card 1/2

67697 sov/126-8-4-21/22 Ye.B., Sukhova, N.A., Arkharov, V.I., Blankova, and Entells, R.A. Investigation of Reaction Diffusion in Binary Systems 18,7530 AUTHORS: PERIODICAL: Fizika metallov i metallovedeniye, 1959, Vol 8, Nr 4, ABSTRACT: Arkharov and Blankova have previously (Ref 1) postulated a correlation between an increase in the relative role of homopolar bonding in diffusion-reaction products and the increase in the relative role of the metalloid in the diffusion process. To check this they have how studied the mechanism of diffusion in the systems:

HE-P, HCO-P, HZr-SA Zr/Se, A Zr/Te/1/Nb-S, Nb-Se and Nb-P, HZr-SA Zr/Se, A Zr/Te/1/Nb-S, the authors These are particularly suited, as the authors explain, to studies of the postulated correlation. results are tabulated, showing for each system the temperature range, the number of macroscopic layers, Nb-Te. number of metallographically distinct layers, phase composition of the scale, whether there is texture in the scale layers and the macroscopic characteristics of Much of the information on scale Card reaction diffusion. 1/2

SOV/126-8-4-21/22

Investigation of Reaction Diffusion in Binary Systems of the Type "Metal-Gas". III

> phase-composition is referred to published work by other From their comparison of data obtained by Various methods the a nors conclude that at 400-600 oc uni-directional diffusion of phosphorus occurs in Fe-P and Co-P; in all the Zr and Nb systems uni-directional diffusion of metalloids is found at 400-1000 °C. results and their comparison with those for other systems confirm that the increase in the relative role of homopolar bonding is associated with an increase in the relative role of the metalloids in diffusion reactions. There are 1 table and 10 references, of which 4 are Soviet, 5 German and 1 Scandinavian.

Card 2/2

ASSOCIATION:

Ural'skiy gosudarstvennyy universitet imeni

A.M. Gorikogo

(Urals State University imeni A.M. Gor'kiy)

SUBMITTED:

June 19, 1959

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sov/126-8-5-11/29

Arkharov, V.I., Klotsman, S.M., and Timofeyev, A.N. AUTHORS:

TITLE:

Card

1/4

On the Effect of Small Additions on the Diffusion

Coefficients in Polycrystalline Materials. III - Effect of Thallium on the Self-Diffusion of Silver

PERIODICAL: Fizika metallov i metallovedeniye, Vol 8, 1959, Nr 5,

pp 709-713 (USSR)

ABSTRACT: Silver (99.99%), remelted in vacuum, and chemically pure

thallium were used as the raw materials in the investigation. Ag-Tl alloys were made in an argon atmosphere. Ingots of pure silver and of the alloy were cold forged into rods of 12 x 12 mm cross-section. All heat-treatment operations and diffusion annealing

were carried out in a vacuum of the order of 10-1 to 10-2 The temperature was maintained with an accuracy

After forging, all specimens were

of t l oc. recrystallized for 6 hours at 900 °C. The grain size of all specimens was practically the same, being approxi-After recrystallization, the specimens of mately 1 mm. pure silver and of the alloy were subjected to one of

the following variants of heat treatment: 1) annealing at 350 °C for 200 hours; 2) annealing at 285 °C for

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On the Effect of Small Additions on the Diffusion Coefficients in Polycrystalline Materials. III - Effect of Thallium on the Self-Diffusion of Silver

200 hours; 3) annealing at 170 °C for 200 hours; and 4) annealing at 170 °C for 200 hours, followed by annealing at 350 °C for 100 hours. Radioactive silver was applied to the surface of the specimens by evaporation in vacuum. All specimens, having undergone one of the above variants of heat treatment, were subjected simultaneously to diffusion annealing. latter was carried out at 285 °C for 200 hours. The distribution of silver in the diffusion zone was determined by a layer analysis using the integral residue method. Measurements of the activity and thickness of the removed layers were carried out employing a method described by Arkharov et al (Refs 1, 2). Figs 1 and 2 are typical graphs for the dependence of the residual integral activity logarithm on the depth of diffusion in pure silver and in the alloy. In a table on page 711, values of diffusion permeability of intercrystalline members of polycrystals of pure silver and a silver alloy containing 0.1% Ti are shown. As can be seen from the

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SOV/126-8-5-11/29

On the Effect of Small Additions on the Diffusion Coefficients in Polycrystalline Materials. III - Effect of Thallium on the Self-Diffusion of Silver

results obtained, the temperatures of maximum adsorption enrichment of the intercrystalline members does not coincide with the temperature of maximum volume solubility. From the independence of the diffusion permeability of the intercrystalline transition zones in pure silver on preliminary heat treatment, it is concluded that the structure of these zones remains unaltered in the investigated temperature range. From data on the dependence of the diffusion permeability of intercrystalline members in the silver - 0.1% Tl alloy on preliminary heat treatment, the adsorption activity of Tl in Ag can be calculated. An experimental confirmation of Arkharov's hypothesis (Ref 8) on the existence of a temperature dependence of the degree of adsorption enrichment was obtained. The sign of the temperature dependence of the intercrystalline internal adsorption of Tl in Ag was derived. It is suggested that the excess energy in the intercrystalline transition zones decreases during adsorption of Tl, mainly due to

Card

On the Effect of Small Additions on the Diffusion Coefficients in Polycrystalline Materials. III - Effect of Thallium on the Self-Diffusion of Silver

its geometrical structure and not due to interatomic reactions.

There are 2 figures, 1 table and 8 references, of which 6 are Soviet and 2 English.

ASSOCIATION: Institut fiziki metallov AN SSSR

(Institute of Physics of Metals, Academy of Sciences

USSR)

SUBMITTED:

May 20, 1959

Card 4/4

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

18.7510

sov/126-8-5-26/29

AUTHORS:

Arkharov, V.I., Borisov, B.S., Vangengeym, S.D., and

Taluts, G.G.

TITLE:

On the Question of the Mechanism of Intercrystalline

Internal Adsorption in Dilute Solid Solutions

PERIODICAL: Fizika metallov i metallovedeniye, Vol 8, 1959, Nr 5,

pp 792-794 (USSR)

ABSTRACT: The interaction between the electron shells of atoms in a dilute solid solution can strongly affect the behaviour of impurity atoms in this solid solution. This applies particularly to intercrystalline internal adsorption. By taking into consideration the electron

interaction it is possible to describe the atomic mechanism of internal adsorption and associate it with quantitative data available in this field. If there are defects or structural non-uniformities in the lattice the impurity atoms react with them. This is a

Card 1/3

long-range order interaction and hence screening must become evident, i.e. the impurity atoms must behave as if they possessed a "screened" atomic radius. structural non-uniformity (among them grain boundaries)

can be considered to be a dislocation system, for an

are musely contained because of the control of the

On the Question of the Mechanism of Intercrystalline Internal Adsorption in Dilute Solid Solutions

approximate description of the interaction between impurity atoms in a solid solution and lattice distortions, it is possible to use Webb's calculation (Ref 11). In this way one can evaluate the number of atoms, No, diffusing through the grain body to the dislocation system modelling the intercrystalline boundary, i.e. the number of atoms experiencing intercrystalline internal adsorption. Such a calculation was carried out by the authors for the solid solutions Ag-Tl, Ag-Zn, Ag-Pb, Cu-Mg and Cu-Sn. The concent of horophilic elements in these alloys were considerably The concentrations lower than their volume solubility. The results of the calculations are reported in the table on p 793. Although the calculated and experimental results agree quite well, a discrepancy can be observed which in a few cases exceeds the absolute errors in lattice parameter measurements. Among the possible reasons for this discrepancy the following can be quoted. First, Webb's formula, which contains macroscopic factors, is somewhat artificial for the description of phenomena of an atomic

Card 2/3

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SOV/126-8-5-26/29

On the Question of the Mechanism of Intercrystalline Internal Adsorption in Dilute Solid Solutions

scale, and can be justified only as a first approximation. Secondly the influence of the relative orientations of neighbouring grains which can change the width of the intercrystalline zone, and the associated lattice parameter (this change varying from one grain group to another) is not taken into consideration. Thirdly the block structure which can change from one test to another can, as a result of internal adsorption at block boundaries, change the magnitude of the lattice parameter somewhat. These facts are subjects for

Card 3/3

further investigation.
There are 1 table and 11 references, of which 4 are

Soviet, 6 English and 1 German.

ASSOCIATION: Ural'skiy gosudarstvennyy universitet: Institut

fiziki metallov AN SSSR (Uralisk State University

(Ural'sk State University: Institute of Physics of

Metals, Acad.Sci. USSR)

SUBMITTED:

June 19, 1959

SOV/126-8-6-10/24

AUTHOR: Arkharov, V.I.

TITLE: On the Question of Non-Uniformities in Concentration

Shown Experimentally to Exist in Solid Bodies 7

PERIODICAL: Fizika metallov i metallovedeniye, 1959, Vol 8, Nr 6,

pp 861-866 (USSR)

ABSTRACT: (Remarks in connection with papers by A.S.Zav'valov and

B.I.Bruk)

The author discusses papers on this subject by Zav'yalov and Bruk (Ref 3 to 7) and disagrees with some of their interpretations. His own conclusions are as follows:

(1) The phenomena of origination of non-uniformity in

concentration in the alloys studied by Zav'yalov and Bruk differ in nature from those of internal adsorption.

(2) The phenomena investigated by Zav'yalov and Bruk can

be explained on the basis of simple thermodynamic circumstances which dictate the form of equilibrium

diagram for the alloy; however, these explanations cannot be applied for internal adsorption phenomena because of the

entirely different nature of the latter.

Card 1/2 (3) The thickness of the intercrystalline internal

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S0V/126-8-6-10/24 On the Question of Non-Uniformities in Concentration Shown Experimentally to Exist in Solid Bodies

adsorption zones which can be estimated from experimental results obtained by different methods, have a magnitude of $10^2 - 10^3$ Å; a direct autoradiographic exposure of such zones is very difficult and is possible only under particularly favourable conditions in particular cases. The zones shown in the papers by Zav'yalov and Bruk have a thickness exceeding the cross-sectional dimensions of intercrystalline internal adsorption zones.

(4) Zav'yalov and Bruk's remarks about certain aspects of the internal adsorption phenomenon contain a number of misunderstandings due to an oversimplified idea about the mechanism of internal adsorption. There are 19 references, 18 of which are Soviet and 1 German.

ASSOCIATION: Institut fiziki metallov AN SSSR (Institute of Metal Physics, AS USSR)

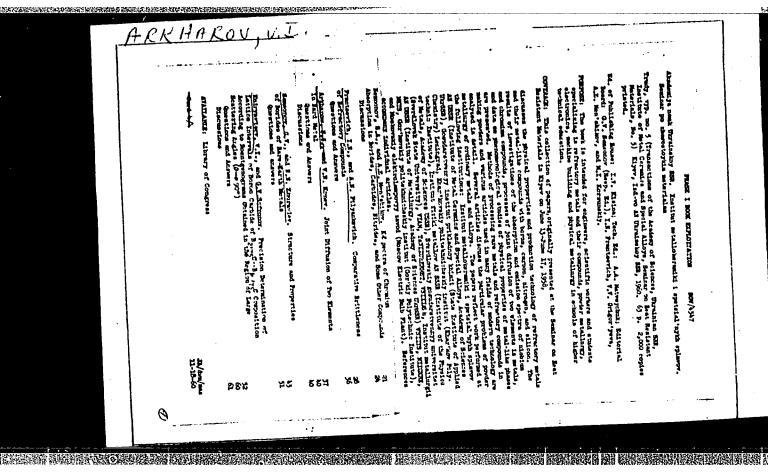
SUBMITTED: July 20, 1959

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"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8



ARKHAROV, U.I

PHASE I BOOK EXPLOITATION

sov/4535

Vsesoyuznyy sovet nauchno-tekhnicheskikh obshchestv

- Mezhkristallitnaya korroziya i korroziya metallov v napryazhennom sostoyanii (Intercrystalline and Stress Corrosion of Metals) Moscow, Mashgiz, 1960. 358 p. 3,000 copies printed.
- Ed.: I.A. Levin, Candidate of Technical Sciences; Ed. of Publishing House: I.I. Lesnichenko, Engineer; Tech. Ed.: V.D. El'kind; Managing Ed. for Literature on Metalworking and Instrument Making (Mashgiz): V.V. Rzhavinskiy, Engineer; Editorial Board: I.A. Levin, Candidate of Technical Sciences (Chairman), V.P. Batrakov, Candidate of Technical Sciences, V.M. Nikiforova, Candidate of Technical Sciences, and A.V. Turkovskaya, Candidate of Technical Sciences.
- FURPOSE: This collection of articles is intended for technical personnel concerned with problems of corrosion of metals.
- COVERAGE: The collection contains discussions of intercrystalline corrosion of stainless steels and stress corrosion of carbon steels, low-alloy and stainless steels, and light-weight and nonferrous alloys. The tendency of steels of CARO I/9

27

S/148/60/000/003/008/018 A161/A029

AUTHORS:

Arkharov, V.I.; Sidorenko, F.A.

TITLE:

Crystallographic Correlations in Thermoelastic Martensite Transformation b

PERIODICAL:

Izvestiya vysshikh uchebnykh zavedeniy. - Chernaya metallurgiya, 1960

A RESERVANTA DESIGNATION AND FLAVOR SHARE SHARE SHARE RESERVANTA AND RESERVANT CONTROL SHARE SHARE AND A SHARE SHA

TEXT: The article presents the contents of a report delivered at the 6th scientific-technical conference on the application of X-rays for investigation of materials. The purpose of the report was to draw attention to certain features of the mechanism of thermoelastic martensite transformation in alloys investigated by G.V. Kurdyumov and L.G. Khandros (15.2% Al, 1.3% Ni, 83.5% Cu). These features were revealed by geometrical calculations based on the orientation correlation of the cubical β 1 and the low-temperature rhombical γ' phases established by Greninger (Ref. 1). The correlation is illustrated by superimposing the two crystallographic lattices (Fig. 1) (the shifts of atoms in transformation are marked by arrows). The transformation process is analyzed and the observed transformation hysteresis is explained by diffusion of atoms "wedged out" from the lattice in

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

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Crystallographic Correlations in Thermoelastic Martensite Transformation

transformation $\beta_1 \rightarrow \gamma^*$. For more detailed explanation of the thermoelastic martensite transformation mechanism the authors determined the heat expansion coefficients of the Kurdyumov-Khandros alloy by roentgenographic means. The statements show the effect of dimensional correlations and their variations with temperature in the transformation process. There are 3 figures and 2 English references.

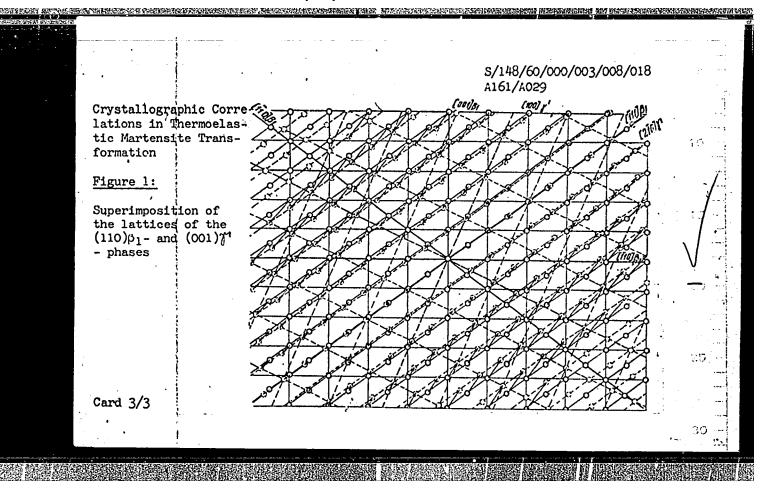
ASSOCIATION: Ural skiy gosudarstvennyy universitet (Ural State University)

SUBMITTED: January 21, 1959

Card 2/3

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8



S/148/60/000/003/013/018 A161/A029

18.7500

AUTHORS:

1555, 1146, 1413

Arkharov, V.I.; Blankova, Ye.B.

Investigation of the Reaction Diffusion in Binary Systems with Tran-

TITLE: ation Flement Base

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. - Chernaya metallurgiya,

1960, No. 3, pp. 122 - 124

The changes of the diffusion mechanism between binary systems with similar properties were studied. A Commercially pure transition metals of the 4th period (Ti/V/Cr/Mn/) Fe/Co/Ni) and oxygen-group metalloids (S/Se/) Te/) were chosen for the study. The interaction of solid metal specimens with vaporized metalloids was studied in the heat interval from the beginning of perceptible scale formation to 1000°C. Macroscopical analysis, microscopic investigation, layerwise roentgenographic analysis of the scale phases and experiments with inert marks were employed. The results led to the assumption that at least in the phases with perceptible metallic properties both the metal and the metalloid affect the quantity of quasi-free electrons, and the metalloid particles in the lattice become positively ionized. The observed fact that in systems with selenum the diffusion of metalloid is strongest in Ti-Se, V-Se and Co-Se, agrees with Card 1/2

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S/148/60/000/003/013/018 A161/A029

Investigation of the Reaction Diffusion in Binary Systems with Transition Element Base

the statement made by Ehrlich (Ref. 3) that the homeopolar and metallic bonds are strongest in these compounds. The following fundamental conclusions were drawn: 1) In binary systems of the 4th-period transitive elements from Ti to Ni, with S, Se and Te for the second component, the diffusion percentage of metal decreases and that of the metalloid increases in the series from sulfides to selenides and then to tellurides. Within the selenide series, the metalloid diffusion is strongest in systems with Ti, V and Co. 2) The easier diffusion of metalloids compared to that of metals is mainly due to homeopolar and metallic bonds in those phases which are the products of reaction diffusion. 3) As the radii of the atoms and of the negative ions of selenium and tellurium are much larger than the radii of the atoms and of the positive ions of metal, it may be supposed that metalloid particles are diffusing through scale (or at least through the scale layers consisting of NiAs and CdJ₂) in the form of positive ions. There are 3 references: 2 Soviet, 1 German.

ASSOCIATION: Ural skiy gosudarstvennyy universitet (Ural State University)

SUBMITTED: January 21, 1959

Card 2/2

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8/137/617000/005/030/060 A006/A106

AUTHORS:

Arkharov, V. I., and Konev, V. N.

THE PROPERTY OF THE PROPERTY O

TITLE:

On the joint diffusion of two elements into solid metal

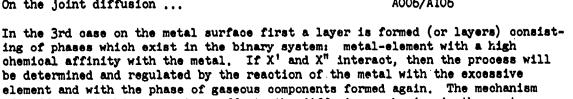
PERIODICAL: Referativnyy zhurnal. Metallurgiya, no. 5, 1961, 2-3, abstract 5Zh15 ("Tr. Seminara po zharostoykim materialam" [In-t metallokeramiki i spets. splavov AN USSR, no. 5] Kiyev, 1960, 37-42)

A theoretical analysis is made of the joint diffusion of 2 elements into a third one in systems of a solid metal and a mixture of two chemically active gases: Me - $(X^{\dagger} + X^{\dagger})$. Phase diagrams of the systems formed by compounds which exist in the binary systems of the Me-X type, are classified. Cases are analyzed where the solubility of X^{\dagger} and X^{n} is unlimited, limited and absent in the binary compounds of the pseudo-binary system $\text{Me}_{n}X^{*}_{m}$ - $\text{Me}_{p}X^{n}_{q}$. In the first case the diffusion layer consists of a single phase with a concentration gradient decreasing along the depth of the layer for both components. In the second case the surface layer consists of a phase with a higher content of the element with least diffusional mobility. At a certain depth a layer of another phase may exist with a higher concentration of elements with a greater diffusional mobility.

Card 1/2

s/137/61/000/005/030/060 A006/A106

On the joint diffusion ...



chemical affinity with the metal. If X' and X" interact, then the process will be determined and regulated by the reaction of the metal with the excessive element and with the phase of gaseous components formed again. The mechanism of diffusion in binary systems affects the diffusion mechanism in the system $Me-(X^* + X^*)$. Some factors are studied which affect the kinetics of the processes. The second case is experimentally confirmed by V. N. Konev and V. I. Arkharov and others (RZhMet, 1959, no. 11, 24702; "Fiz. metallov i metallovedeniye" 1952, v. 5, 192). The first and third case were not specially studied.

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[Abstracter's note: Complete translation]

Card 2/2

CIA-RDP86-00513R000102110008-8 "APPROVED FOR RELEASE: 06/05/2000

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S/137/62/000/004/078/201 A052/A101

AUTHOR:

Arkharov, V. I.

TITLE:

Non-uniformity of component distribution in solid solutions

PERIODICAL:

Referativnyy zhurnal, Metallurgiya, no. 4, 1962, 5, abstract 4135

("Tr. In-ta fiz. metallov, AN SSSR", no. 23, 1960, 87 - 102)

A review of the results of experimental investigations of internal adsorption in solids. A conclusion is drawn that in a real solid solution the TEXT: distribution of admixtures must be, as a rule, non-uniform, and that a non-uniformity of distribution of concentration of adsorption origin in a number of cases is of a first class importance for the properties of a solid as a whole. A particularly strong effect on the properties of a solid has the internal adsorption of some small admixtures which, when their adsorption is eliminated or prevented, change inconsiderably the properties of a solid.

A. Rusakov

[Abstracter's note: Complete translation]

Card 1/1

<u>arterandur perestipuntan pertaturan dan pertatuk pertatu</u>

68483 \$/126/60/009/01/015/031 18.7500 E111/E191 Arkharov, V.1., Borisov, B.S., Vangengeym, S.D., and 18.8100 AUTHORS: Sokolova, G.K. Investigation of the Connection between Internal Adsorption in Alleys and their Electrical Resistance TITLE: PERIODICAL: Fizika metallov 1 metallovedeniya, 1960, Vol 9, Nr 1, pp 81~85 (USSR) ABSTRACT: The authors discuss methods which have been used for the experimental study of internal adsorption, noting the general difficulties of such investigations. Their own previous work (Refs 1-3) in which changes in lattice parameters associated with adsorption effects in many

Card

parameters associated with adolphisms. Was followed by binary and ternary alloys were studied, was followed by an attempt to provide a qualitative explanation (Ref 4). The arther work has shown that neither the thermodynamic nor a qualitative molecular-kinetic approach is sufficient to a qualitative mechanism of inter-crystallite internal elucidate the mechanism of inter-crystallite internal adsorption. The authors show that internal adsorption, being related to the ionic sub-lattice of the crystal, must be closely connected with electronic processes in the dilute solid-solution lattice. They go on to

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S/126/60/009/01/015/031 E111/E191

Investigation of the Connection between Internal Adsorption in Alloys and their Electrical Resistance

consider the electrical resistance of dilute solidsolutions and then to describe experiments aimed at checking the model ideas by resistance determinations on such solutions relative to the grain size of polycrystalline specimens. The alloys studied were Cu - 1% Cd, Cu - 1% In, Cu - 0.6% Sn, and Cu - 1% Sb, made from electrolytic copper (99.99% Cu) and granulated (99.9%) other elements. Alloys were melted in quartz crucibles under borax and after forging and homogenizing armealing were out longitudinally, each half being drawn into a wire 410 mm long and 0.3 mm in diameter. Resistance was determined at temperatures from -192 to +25 °C directly after drawing (curves 1 in a figure on p 84, showing resistance as a function of temperature) and after low-temperature (400-600 °C) tempering (curves 2) and high-temperature (800-900 oc) temporing followed by slow cooling (curves 3). Resistance for all alloys increased with temperature, decreased somewhat at any given temperature on low-

Card 2/4

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

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\$/125/60/009/01/015/031 E111/E191

Investigation of the Connection between Internal Adsorption in Alloys and their Electrical Resistance

ASSOCIATION: Ural skiy gosudarstvennyy universitet imeni

A.M. Ger'kogo (Ural'sk State University imeni A.M. Gor'kiy)

SUBMITTED:

October 9. 1959

Card 4/4

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APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

\$\126\60\009\02\010\033

AUTHORS: Arkharov, V.I. and Konev, V.N. E111/E335

TITLE: Investigation of Reaction Diffusion in "Metal-complex

Gas" Systems. I. General Picture of the Phenomenon

PERIODICAL: Fizika metallov i metallovedeniye, 1960, Vol 9, Nr 2,

pp 212 - 215 (USSR)

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The authors point out that to be useful in technology, ABSTRACT:

studies of diffusion with reaction should involve mixtures of gases as well as of solid phases. The general aim is to study such diffusion (especially gas corrosion and processes for making high-temperature coatings) and find practical methods of regulating these processes: for this many particular cases of systems of the type "solid element - mixture of two chemically active gaseous components" can be formulated. The authors discuss the significance of the type of equilibrium diagram of binary and ternary systems of the elements participating in diffusion with reaction. They examine three main types of pseudobinary systems in which reaction

of the three components forms a scale with a phase Card1/2

composition corresponding to variations in the character

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

S/126/60/009/02/010/033 Investigation of Reaction Diffusion in #111/E335 Metal-complex Gas" Systems. I. General Picture of the Phenomenon

of pseudobinary equilibrium diagrams formed by chemical compounds contained in binary systems of the metal with each of the components separately. The authors also consider briefly the significance of the type of crystal structure of phases formed in the scale and finally indicate the possible importance of other factors, such as phase transformations in the metal during solution in it of the gaseous elements or formation of deposits on the solid surface through gas-phase reactions.

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ASSOCIATION: Ural'skiy gosudarstvennyy universitet im. A.M. Gor'kogo (Ural State University imeni A.M. Gor'kiy)

SUBMITTED: November 2, 1959

Card 2/2

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

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THE REPORT OF THE PROPERTY OF

Konev, V.N., Bogacheva, N.G. and Arkharov, V.I.

AUTHORS:

Investigation of Diffusion with Reaction in the System

TITLE:

"Metal-Complex Gas". II. The System Chromium - 1

Sulphur-Dioxide A

PERIODICAL: Fizika metallov i metallovedeniye, 1960, Vol 9, Nr 3,

pp 358-361 (USSR)

ABSTRACT:

This is a continuation of work by Arkharov and Konev (Ref 1) to investigate the physical mechanism of diffusion with reaction in systems of the type solidmetal-mixture of chemically active gases. This information is necessary for developing non-scaling materials and understanding their failures in service. The present work deals with Cr-0-S. Parallelepiped specimens (1.0 to 1.5 cm side) of technical chromium were suspended in a heated sealed quartz tube at 600 to 1200°C; the apparatus and procedure were described previously (Ref 2 to 6). Fig 1 shows gains in weight of specimens at the various temperatures (except 600 and 1200°C) per unit of surface as functions of time (hours). Fig 2 shows a plot of the logarithm of the rate constant

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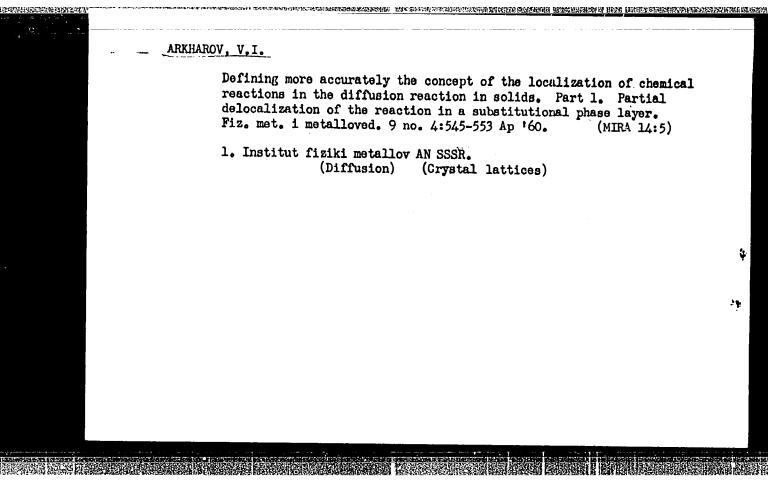
Investigation of Diffusion with Reaction in the System "Metal-Complex Gas". II. The System Chromium - Sulphur-Dioxide

against reciprocal of absolute temperature, the points relating to those in Fig 1, where the curves become horizontal. Specimens exposed under various conditions were subjected to qualitative X-ray analysis: a feature was the appearance of a new phase. Fig 3 shows lines from the outside (a) and inner (6) layers of a specimen oxidized in sulphur dioxide at 1200°C, some doubling of lines being evident in the latter. No texture in the outer layer of scale could be detected on any specimen. The scale contained 9.23 weight % of combined sulphur, according to analyses carried out in the Mineral Salts Heating of mixtures of Cr203 and Laboratory of UNIKhIM. CrS, or Cr203 and Cr2S3 at 800°C for 8 to 10 hours in vacuo (table gives compositions of mixtures and products) did not give the new phase present in the scale. The non-scaling properties of specimens previously treated under conditions producing the new phase were suspended in air at 1000 and 1100°C: in 12 hours no weight increase took place and the new phase remained. The investigation

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80529 S/126/60/009/05/008/025

AUTHORS:

Arkharov, V.I., Konev, V.N. and Gerasimov, A.F.

TITLE:

Investigation of Diffusion the Reaction in "Metal - Complex Gas" Systems. iv. The System Molybdenum-nitrogen-carbon

PERIODICAL:

Fizika metallov i metallovedeniye, 1960, Vol 9, Nr 5, pp 695 - 700 (USSR)

ABSTRACT:

This reports a further investigation by this school on diffusion with reaction in systems of the "metal - mixture of two chemically-active gases" type (Refs 1-7). V. Negodyayev and G. Tatymov participated in the experiments, which were carried out on the binary molybdenum-nitrogen (method described in Refs 15, 16) and molybdenum-carbon systems (method described in Refs 11,12) and then on the ternary molybdenum-nitrogen-carbon system (method described in Refs 3,4). Figure 1 shows increases in weight of molybdenum specimens in ammonia vapour as functions of time for 700 - 1 120 °C. Figure 3 the corresponding curves for a benzene-hydrogen atmosphere at 1 000 - 1 200 °C, and Figure 4 for a benzene-ammonia atmosphere at 1 000 - 1 200 °C. The dependence of the

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Investigation of Diffusion with Reaction in Metal - Complex Gas" IV. The System Molybdenum-nitrogen-carbon Systems.

logarithm of the parabolic constant on reciprocal of absolute temperature is shown for all the systems in Figure 2 and the values of the constant are given in Table 2. The lines consist of thrue straight sections and the authors give a physical interpretation of this. Table 1 gives for all the systems the number of layers detected metallographically, the results of phase X-ray analyses and the presence or absence of texture at the various temperature ranges. Texture was found only in Mo-C at 1 200 °C. The work showed that molybdenumnitrogen reaction in an ammonia atmosphere proceeds appreciably at 700 °C following the parabolic law up to 150 °C, above which molybdenum nitrides were not formed. At 700 - 940 °C a scale of an outer layer of MoN with a simple hexagonal lattice and an inner layer of ${
m Mo}_2{
m N}$

with a face-centred cubic lattice was produced. There was no reaction between molybdenum and molecular nitrogen at atmospheric pressure and 600 - 1 200 °C. The reaction with carbon (from benzene + hydrogen) proceeded appreciably at

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Pastati dipita isto

80530 18.7530 S/126/60/009/05/009/025 AUTHORS: Konev, V.N. and Pavlova, V.P. TITLE: Investigation of Diffusion with Reaction in "Metal -Complex Gas" Systems. V. The System Chromium Sulphurnitrogen/ PERIODICAL: Fizika metallov i metallovedeni e, 1960, Vol 9, Nr 5, pp 701 - 708 (USSR) ABSTRACT: This is a further contribution to the series of researches by this school on diffusion with reaction in systems of the "metal - mixture of two chemically-

active gases" type (Refs 1-7). In the work the authors extended their previous experiments on the chromium—sulphur system (Ref 10) before proceeding to the ternary parallelipiped) specimens of electrolytic chromium were suspended by quartz in a furnace (Figure 1). For the evacuated and its lower end was kept at 250 °C to give a system the tube after evacuation was connected to

system the tube after evacuation was connected to a source Cardl/3 of pure nitrogen. The products were examined as described

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Investigation of Diffusion with Reaction in Metal - Complex Gas"

Systems. V. The System Chromium-sulphur-nitrogen

previously (Refs 1,8,9,12). The experiments were carried out at 700, 800, 900 and 1 000 °C at a constant sulphur partial pressure; nitrogen pressure was constant at 25 to 30 mm Hg (gauge). Table 1 gives the number of layers detected metallographically, the results of X-ray phase analysis, presence or absence of texture and the microscopic characteristics of the diffusion mechanism. Figure 2 shows typical appearance of a specimen initially and after treatment at 1 000 and 700 °C. Figure 3 shows a section through a specimen sulphided at 1 000 °C for 1 hour and Figures 4a and 4b one through a specimen treated with sulphur + nitrogen for 4 hours at 1 000 and 2 hours at 700 °C, respectively. The weight-gains of specimens under the various conditions are shown as functions of time in Figure 5 and the logarithm of the parabolic constant of the rate curves as functions of reciprocal of absolute temperature in Figure 6 (linear for the binary, complex for the ternary). Reaction diffusion in both systems Card2/3 follows the parabolic law for 700 - 1000 °C (constant

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Investigation of Diffusion with Reaction in Metal - Complex Gas" Systems. V. The System Chromium-sulphur-mitrogen

values are given in Table 2). The rate of scaling of the binary is greater than that of the ternary system. In the binary system the outer light layer approximates to Cr_2S_3 at 1 000 °C and Cr_3S_4 at 700 °C, while the dark inner layer approximates to Cr_5S_6 . The work showed that in this system the reaction diffusion involves movement of the components in both directions through the scale layer, the relative importance of chromium growing with increasing temperature. In the ternary system the process depends on diffusion of sulphur and nitrogen through crystal lattices to the metal and of chromium to the scale outer surface. There are 6 figures, 2 tables and 16 references, 13 of which are Soviet, 2 German and 1 international.

ASSOCIATION: Ural'skiy gosudarstvennyy universitet im.
A.M. Gor'kogo (Ural State University imeni A.M. Gor'kiy)

SUBMITTED: December 23, 1959

Card3/3

TO RELIGIOUS PROPERTY AND CONTRACT AND CONTRACT AND AND AND CONTRACT AND CONTRACT AND AND CONTRACT AND CONTRA s/126/60/009/05/010/025 Closer Definition of the Ideas on Localization of Chemical Reaction During Diffusion With Reaction in Solids.
The Conditions for the Most Complete Conditions for the Most Conditio Reaction puring <u>pullusion</u> Neaction in Solids.

II. Conditions for the Most Complete Localization of Conditions for the North Complete Localization of Conditions of Cond 11. Conditions for the Most Complete Localization of Reaction at the Boundaries of a Layer of a Substitution-AUTHOR: Fizika metallov i metallovedeniye, 1960, Vol 9, Nr 5, TITLE: ABSTRACT: The author gives a theoretical treatment of concentration the greatest distribution through a diffusion layer giving the greatest The author gives a theoretical treatment of concentration distribution through a diffusion layer giving the greatest distribution through a dillusion layer giving the greatest localization of chemical reactions at the layer boundaries. The type of distribution previously shown by him to be PERIODICAL: The type of distribution previously shown by him required for a solid solution based on a Memon is shown in Figures la and lb, relating, respectively, to localization at the lefthand or righthand boundary.

Rights la 311 Mo of one of soon as they are at the righthand Figure 1a all Me atoms, as soon as they are at the righthand side of the lefthand houndary. rigure is all me atoms, as soon as they are at the righthan side of the lefthand boundary, are practically completely combined with O stoms arriving by diffusion through the combined with 0 atoms arriving by diffusion through the mass to form the compound with maximum excess of Me over the stoichicmetric compand mass to form the compound with maximum excess of me over the boundary the card1/3 the stoichiometric compound; at the other boundary,

S/126/60/009/05/010/025

Closer Definition of the Ideas on Localization of Chemical Reaction During Diffusion with Reaction in Solids. II. Conditions for the Most Complete Localization of Reaction at the Boundaries of a Layer of a Substitution-type Phase

concentration of Me falls to the lowest level corresponding to equilibrium of the compound with pure 0. In Figure 1b the rôles of Me and O are reversed. Using his former notation the author deduces diffusion equations for the above cases. When reaction is localized simultaneously at both boundaries concentration distribution is represented by a combination of Figures la and lb. The concentration distribution represented by these figures does not necessarily secure localization at one side. intensive diffusion of Me and of O their atoms can penetrate beyond the limits shown. The narrow zone where the concentration gradient is steep is formed in the earliest stage of reaction diffusion. The opposite sides of this phase represent composition deviating from the stoichiometric compound to coexist with pure Me and pure 0, respectively. The possible forms of its further growth are shown in the series of Figure 2, concentration

Card2/3

5/126/60/009/06/011/025

Arkharov, V.I. and Blankova, Ye.B. 52 AUTHORS:

Structural Characteristics of Scale Which are Used in the Investigation of the Mechanism of Diffusion with Reaction TITLE:

PERIODICAL: Fizika metallov i metallovedeniye, 1960, Vol 9, Nr 6,

pp 878 - 887 (USSR)

The authors develop and give a more detailed treatment ABSTRACT:

of the ideas published by one of them (Arkharov -Refs 1, 2) on the nature of the participation by components reacting with diffusion. They give the following as the cardinal questions. Do atoms diffuse as ions of "normal" or "average" valency or as neutral atoms with "normal" or "screened" radius: Is compoent movement effected through

interstices or vacancies and do the atoms of a given component use only its own sublattice? The answers are obtainable from examinations of scale. Macrostructural studies are particularly fruitful but can become complicated.

The authors discuss structural indications used for evaluating the degree of participation of components in the

diffusion process. Texture in the scale layers is important, an increase in the growth texture indicating

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S/126/60/009/06/01/025 Structural Characteristics of Scale which are Used in the Investigation of the Mechanism of Diffusion with Reaction

under otherwise equal conditions, an intensified role of the given component in the overall diffusion process; but in their discussion of this effect the authors show that this is not unambiguous since recrystallization texture can also arise, and point to other possible complications. They emphasise that all structural characteristics in all scale layers should be examined. Grain/size distribution also gives valuable indications (e.g. the presence in the scale layer of a finer-grain zone at the scale/solid boundary suggests diffusion of a component from the opposite scale boundary) but here, too complicating factors exist. Further evidence can sometimes be obtained from changes in the form of external scale surface, especially when one component is liquid or gaseous and the solid specimen is initially a rectangular parallelipiped: one-way diffusion of one component is detectable but with two-way diffusion of both the visible effects may disappear. Pfeil's (Ref 11) inert-marker method is useful if the relative size of markers is sufficiently small but their fixing is

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Structural Characteristics of Scale Which are Used in the Investigation of the Mechanism of Diffusion with Reaction

difficult. Although each of the methods discussed gives only qualitative indications, the authors show that their suitable combination enables primary and secondary effects to be distinguished. They give a tabulation of these (Table 1) and note that further indications can be obtained from the concentration gradient, whose determination is, however, often difficult. Table 2 gives a scheme for grading observed effects to obtain the most reliable estimate of the role of each component in the process. There are 2 tables and 12 references, 9 of which are Soviet, 2 English and 1 German.

ASSOCIATION: Ural'skiy gosudarstvennyy universitet im. A.M. Gor'kogo (Ural State University imeni A.M. Gor'kiy)

SUBMITTED: December 26, 1959

Card 3/3

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

ARKHAROV, V.L.

S/126/60/010/01/007/019 E111/E335

Arkharov, V.I. and Blankova, Ye.B. AUTHORS:

The Degree of Participation of Components in Diffusion

with Reaction in High-temperature Oxidation of TITLE:

Certain Transition Metals

Fizika metallov it metallovedeniye, 1960, Vol.10, PERIODICAL:

No. 1, pp 63 - 69

The authors have previously studied diffusion in a number of binary systems in which one component is a fourth or fifth-group transition metal and the other is either sulphur, selenium or tellurium (Refs.1-3). To obtain a fuller and more correct background on the mechanism of the high-temperature oxidation of metals the authors now relate, using mainly published but some original data, these results to those on similar systems in which oxygen participates in place of sulphur, etc. To evaluate the degree of participation of components in diffusion they use their previously described system (Ref. 4). Published data for titanium (Refs5-10) show that on their system (Ref,4) the mechanism differs depending on whether the temperature is above or below 900°C; at higher temperatures VCard1/4

S/126/60/010/01/007/019 E111/E335

The Degree of Participation of Components in Diffusion with Reaction in High-temperature Oxidation of Certain Transition Metals

the role of metal diffusion increases. Varidium (Ref.11) has a first stage of scaling (400 - 600 C in air) in which oxygen diffusion and reaction occurs in the metal/scale interface; later metal diffuses. Previous conclusions (Ref. 12) on diffusion in chromium oxidation require revision in the light of later work (Ref. 13); there are indications, however, of oxygen diffusion, although data are insufficient for firm conclusions. With manganese diffusion is in both directions (Ref. 16). The authors describe additional experiments with inert markers (Figure 1) which suggest that manganese diffusion plays the greater part. With iron, Noxygen diffusion predominates below 400 °C; at higher temperatures both elements diffuse and the scale has a complex character (Refs. 18-20). The authors report additional experiments with cobalt (Fig. 2 shows appearance of a specimen with platinum wire on its surface, oxidised in air for 8 hours at 1000 °C): these confirm previous indications (Refs 20, 21) that intensive diffusion of oxygen Card 2/4

S/126/60/010/01/007/019 E111/E335

The Degree of Participation of Components in Diffusion with Reaction in High-temperature Oxidation of Certain Transition

occurs. Dealing with nickel the authors discuss divergent published results (Refs. 20-24) and suggest an explanation which is contrary to calculations based on Wagner's theory. Oxidation of zirconium, niobium and molybdenum and tungsten occurs by diffusion of oxygen. Finally, the authors deal with two non-diffusion metals, copper and zinc. For the former additional transition metals, shows an inert marker in a fully oxidized experiments (Fig. 3 shows an inert marker in a fully oxidized experimen) supported the view (Refs. 27-29) that copper-ion specimen) supported the view (Refs. 27-29) that copper-ion diffusion is mainly responsible and not two-directional diffusion (as suggested in Refs. 30 and 31). Zinc is oxidised with (as suggested in Refs. 30 and 31). For both copper and zinc diffusion of the metal (Ref. 32). For both copper and zinc the mechanisms are in good agreement with Wagner's theory. The authors assign the type of diffusion for each of the transition authors assign the classification proposed previously (Ref. 4).

card 3/4

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The Degree of Participation of Components in Diffusion with Reaction in High-temperature Oxidation of Certain Transitions

There are 3 figures and 32 references: 15 Soviet, 13 English, 2 German. 1 Czech and 1 Scandinavian.

ASSOCIATION:

Ural'skiy gosudarstvennyy universitet im. A.M. Gor'kogo (Ural State University im. A.M. Gor'kiy)

SUBMITTED:

March 21, 1960

Card 4/4

ARKHAROV, V.I.

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E111/E352 Blankova, Ye.B.

AUTHORS:

Arkharov, V.I. and Blankova, Ye.B.

TITLE:

Investigation of Diffusion with Reaction in Binary

Systems of the Type "Metal-Gas". IV.

PERIODICAL: Fizika metallov i metallovedeniye, 1960, Vol. 10, No. 2, pp. 226 - 232

TEXT: The authors continue their series of investigations (Refs. 1, 2) aimed at elucidating the factors determining diffusion mechanism in metal-gas systems. They review their previous conclusions and analyse further experimental data. They conclude from this and general theoretical considerations that the main crystallographic factor is the nature of the bonding forces between the components in the reaction-product crystal lattice. In systems "transition metal-metalloid (or gas)" an increase in the relative importance of homopolar bonds in compounds can be related to increase in the relative role of diffusion of the metalloid in the overall reaction diffusion process. They suggest that an elementary diffusion function is possible in which, because of existence of homopolar bonds, a local redistribution of electron density occurs with formation Card 1/2

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Investigation of Diffusion with Reaction in Binary Systems of the Type "Metal-Gas". IV.

of an atomic complex including a metalloid ion located in the vacancy of the metal sub-lattice, and its closest surrounding ions of metal with locally charged valency. This provides an explanation for the experimentally observed change in the relative role of diffusion of the components when the composition of binary diffusion systems is varied, particularly in those cases which are not explicable on purely crystallogeometrical considerations. On the basis of these ideas the authors examine and explain reaction-diffusion mechanism in a number of binary systems, including titanium vanadium, chromium, manganese, iron, cobalt | nickel , copper, Nzinc , zirconium, nniobium , molybdenum and tungsten with oxygen sulphur, Mselenium for tellurium; A also in Fe-P, Co-P and Ni-P. There are 1 table and 16 references: 10 Soviet, 3 English, 2 German and 1 international.

ASSOCIATION:

Ural'skiy gosudarstvennyy universitet im.

A.M. Gor'kogo (Ural State University im.

A.M. Gor'kiy) March 21, 1960

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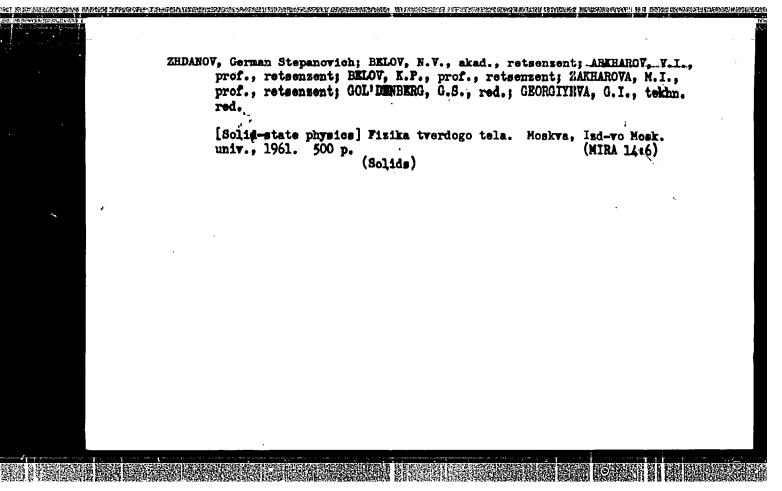
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ARKHAROV, V.I.; BORISOV, B.S.; VANGENGEYM, S.D.

Revealing the characteristics of crystallite interlinking in an I-ray scattering image. Fiz. met. i metalloved. 10 no.3:367-374 S'60. (MIRA 13:10)

1. Institut fiziki metallov AN SSSR i Ural'skiy gosudarstvennyy universitet im. A.M.Gor'kogo. (Crystal lattices) (X rays--Scattering)

"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8



S/659/61/007/000/024/044 D217/D303

18.1731 AUTHORS:

Arkharov, V.I., and Konev, V.N.

TITLE:

Investigating the diffusion kinetics for reaction within systems consisting of metals with complex vapors

SOURCE:

Akademiya nauk SSSR. Institut metallurgii. Issledovaniya po zharoprochnym splavam, v. 7, 1961, 221 - 226

TEXT: An analysis of various structures found in the diffusion layers of diffusion-reaction products, within systems consisting of metals in the presence of two chemically active gases, was carried out. Structural pictures of diffusion-reactions were obtained for the systems Cr-O, Cr-C, Cr-N, Cr-S, Cr-S-N, Cr-S-O, Cr-N-O and Cr-N-C. The structural pictures obtained were tied up with general theories on the reaction mechanism of complex systems of the above type. It was established that it is possible in principle to use diffusion-reaction layers in the systems Cr-C, Cr-C-N, Cr-O-S as protective coats. There are 1 figure, 1 table and 4 Soviet-bloc references.

Card 1/1

18.9200 1145, 1418, 1555

\$/126/61/011/003/005/017 E193/E483

AUTHORS:

Arkharov, V.I., Borisov, B.S. and Vangengeym, S.D.

TITLE:

Manifestation of the Process of Internal Intergranular

Adsorption in the X-Ray Scattering Pattern

PERIODICAL: Fizika metallov i metallovedeniye, 1961, Vol.11, No.3,

It was shown earlier by the present authors (Ref. 8) that TEXT: the phon intensity of X-ray diffraction patterns, obtained on massive polycrystalline specimens, is higher than that of patterns produced by powder specimens of the same material with grains of the same size. This effect, attributed to the presence of an amorphous layer in the polycrystalline specimens, was much less pronounced in pure materials which indicated preferential adsorption of the impurity atoms in the grain boundary regions. The data, then obtained, were insufficient to form any conclusive opinion regarding the behaviour of impurities in the region of intercrystalline internal adsorption and to determine whether these impurities are present in the solid solution or in the second phase; hence, the investigation described in the As before, the experiments were carried out on Card 1/4

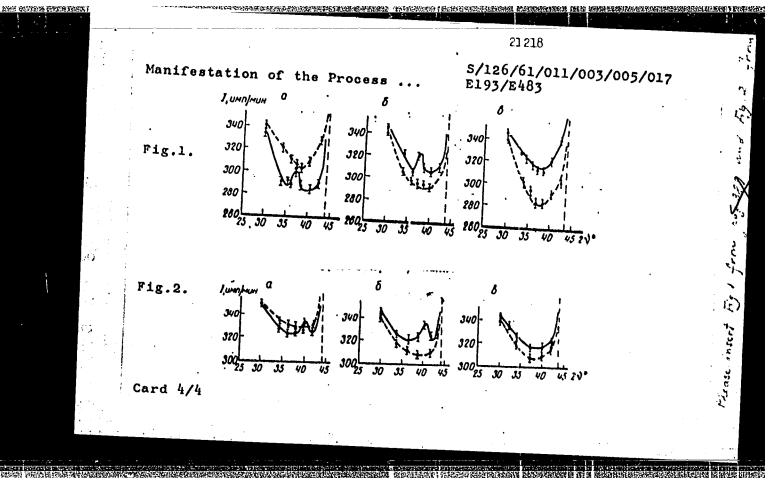
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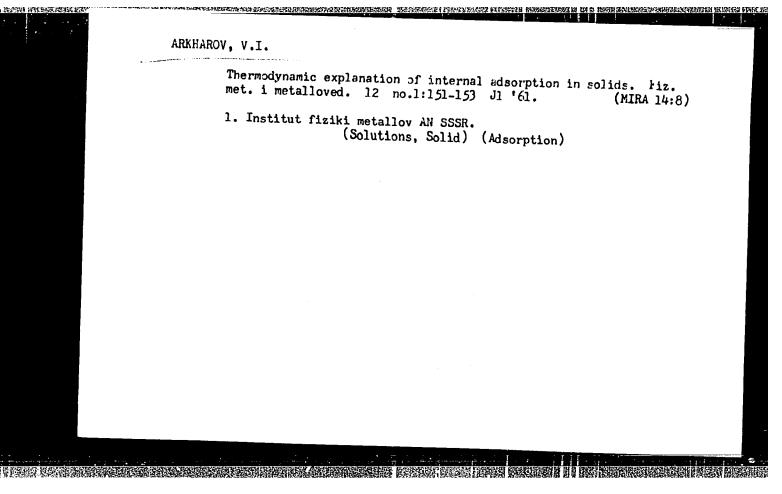
both massive and powder specimens, characterized by the same grain size. The experimental materials consisted of electrolytic copper (99.94% purity) and a Cu-base alloy containing 0.35 wt.% antimony. The results of the measurements of the intensity of X-ray scattering are reproduced graphically. In Fig.1, the intensity of the scattered X-ray beam (I, pulses/min) is plotted against the diffraction angle 20 for (a) plastically deformed specimens, (b) specimens annealed for 2h at 400°C and (B) specimens annealed for 6h at 800°C, the continuous and broken curves relating respectively to massive and powder Cu specimens. The results obtained for the Cu-Sb specimens are presented in the same manner in Fig.2, except that graphs (b) and (B) relate respectively to specimens annealed for 2h at 450°C and for 6h at It will be seen that whereas the I(29) curves for the massive polycrystalline specimens of both Cu and Cu-Sb alloy showed intensity peaks (shifted for the (111) lines of Cu towards the smaller values of 2%), no intensity peaks were observed on curves constructed for the powder specimens. Since no peaks were observed on the I(2) curves for massive specimens of high (99.999%) purity copper, it was concluded that these peaks are associated with the

21218 S/126/61/011/003/005/017 Manifestation of the Process ... E193/E483 formation of a small quantity of a new phase, precipitated in the region of intercrystalline internal adsorption from the locally supersaturated solid solution. In the case of electrolytic copper in which tin was the main impurity, the appearance of the X-ray scattering intensity peak was caused by the formation of the Cu3Sn phase; the peak on curves, constructed for the Cu-Sb alloy, was due to the precipitation of the Cu3Sb phase. There are 2 figures and 9 references: 7 Soviet and 2 non-Soviet. ASSOCIATIONS: Institut fiziki metallov AN SSSR (Institute of Physics of Metals AS USSR) Ural'skiy gosudarstvennyy universitet im. A.M.Gor'kogo (Ural State University imeni A.M.Gor'kiy) SUBMITTED: October 3, 1960 Card 3/4



APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8



S/126/61/012/002/007/019 E202/E335

AUTHOR:

Arkharov, V.T.

TITLE:

Problems of Modelling the Structure of Inter-

THE CONTROL OF THE PROPERTY OF

crystallite Bonding

PERIODICAL: Fizika metallov i metallovedeniye, 1961, Vol. 12,

No. 2, pp. 223 - 232

TEXT: A general method of elucidating the structure of the intercrystallite bonding is described. This method an be looked upon as an extension and considerable generalisation of the Hargreaves and Hill (Ref. 2 - J'. Inst. Netals, 1929, 41, 269) model. The new model is extended to three dimensions and is applicable to any type and angle of dislocation. The author assumes that regardless of the thickness of the intercrystallite transition zone, the position of the atoms in the central part of the transition zone is affected equally by the two differently orientated lattices meeting at the boundaries of the two respective crystallites. Hence, the position of these central atoms should be intermediate between that which they would have occupied, were they the peripheral atoms of one of the crystallites in the absence of the other and vice

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Problems of Modelling

S/126/61/012/002/007/019 E202/E335

This intermediate position is found by taking the projections of each of the adjacent crystallites which in a given disorientation lie nearest to the plane parallel to the plane of the boundary. In this way a system of double configuration is plotted, comprising point intersections of the two projections, which forms a kind of two-dimensional raster. This model is extended further by introducing a threedimensional concept, viz. a packet of crystallographic planes which contains a known number of intersections within the boundary of a right parallelipiped. It is the latter which can he looked upon as a unit cell of the intercrystallite bonding structure modified by the mutual interaction of the differently orientated regions. For each crystallite bonding, the number of planes in a packet is chosen in such a way that it is equal to the number of intersections (i.e. the number of atoms) in the elementary unit in both of the packets compared, and the packets chosen in this manner are defined as e q u i v a l e n t packets. The ultimate structural model

Card 2/6

Problems of Modelling

S/126/61/012/002/007/019 E202/E335

is based on the supposition that a number of successive equivalent packets is positioned along the normal to the plane of the boundary. By taking two equivalent packets in a given disorientated intercrystallite bonding, correlating their thickness and then superimposing them in such a way as to make the vertices and sides of the elementary parallelipipeds coincide, a three-dimensional "grid" is obtained. On it, the lines Joining the two adjacent points of intersection are the lines of dislocation by means of which the crystallographic lattice of one orientation can be transformed in the simplest way into a lattice of the same crystallographic type but different orientation. Consequently, the midpoints of these lines will correspond to the positions of the atoms in the centre of the transitional zone between the two crystallites. The above argument is applied to the study of the intercrystallite zone by superimposing two crystallographic projections, viz. (100) and (111) of a primitive subject lattice, with the orientation [001] parallel to [110]. This yields an elementary rectangular loop 7a x 5a Next, knowing the reticular density for (100) and (111) the minimum number of

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Problems of Modelling

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planes in the corresponding equivalent packets is found to be 4 and 7, respectively, and the stress corrected number of atoms per unit surface of the projection in each packet - 4. Similarly, the total number of atoms in each of the elementary parallelipipeds is 140. In packet (.00) each of the four planes contains 35 atoms, while in (ill) each of the seven planes contains 20 atoms. Hence, in order to transform the elementary parallelipiped, the seven planes of (111) must transform into the four planes of (100), and this can be only achieved by shifting the atoms from layers II, IV and VI, of (111) into the last four layers so as to bring the total in each layer to 35 atoms. This is illustrated in Fig. 2, which shows the schematic cross-section of the intercrystallite bonding, at a right-angle to the plane of the boundary. Now, it is possible to draw the positions of atoms in each of the layers and thus elucidate the whole atructure of the inter-

Card 4/6

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Problems of modelling

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E202/E335

There are 6 figures and 13 references: 5 Soviet and 8 non-Soviet. The four latest English-language references quoted are: Ref. 3 - J.M. Burgers - Proc. Phys. Soc., 1940, Ref. 5 - W.T. Read, W. Shockley - Phys. Rev., 1950, 78, 275; Ref. 6 - N.F. Mott - Proc. Phys. Soc., 1948, 60, 391

ASSOCIATION: Institut fiziki metallov AN SSSR (Institute of Physics of Metals of the AS USSR)

SUBMITTED: December 20, 1960

Card 5/6

38659 S/126/61/012/005/019/028 E073/E535

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AUTHORS:

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Arkharov, V.I., Gerasimov, A.F. and Ushkova, T.V.

TITLE:

On high temperature oxidation of niobium

PERIODICAL: Fizika metallov i metallovedeniye, v.12, no.5, 1961,

761-763

The authors investigated the oxidation of metallic niobium in air in the temperature range 700 to 1200°C. The kinetic measurements were made by the method of continuous weighing; the phase composition of the products of oxidation were determined by means of X-rays, with Fc-Kα-radiation aby the Debye method The kinetic measurements using Cylindrical and flat specimens. confirmed the results obtained earlier by W. Klopp, C.T. Sims and R. J. Jaffee (Ref.4: J. Trans. ASM, 1959, 51, 282) on the linear increase of scale and the anomaly of the oxidation speed at 800°C. According to Klopp et al., this anomaly may be due to the geometry of the specimen, to differences in the air humidity from one test to another and to sintering of the scale at high tempera-The authors of this paper studied the kinetics of oxidation on specimens of various shape, applying various air

Card 1/2

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

S/126/61/012/006/008/023 E132/E135

Arkharov, V.I. AUTHOR:

Regions of coherent reconstruction of the crystal lattice in phase transformations in solid bodies TITLE:

PERIODICAL: Fizika metallov i metallovedeniye, v.12, no.6, 1961, 853-859

Methods are given for calculating the form and dimensions to be expected for coherent regions of a new solid TEXT: phase produced by a solid-state transformation from an initial solid phase. Starting from a certain atom which is assumed to be unchanged in position by the transformation, the displacements of other atoms increase with distance from the first, depending on the relationship between the two structures, until the displacement approaches half a unit cell dimension. it may be easier for successive atoms to move in the opposite direction from a new centre. The size of the particles generated is thus limited and is expected to be larger when the displacements giving the new phase are smaller.

Card 1/2

这是是我的时代,他们是这种关键,他们的对象所有对象的特别,我们就是他的对象的的对象的对象的。

Regions of coherent reconstruction... \$\frac{5}{126}\frac{61}{012}\frac{1006}{008}\frac{023}{023}

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Theoretical examples of the calculation are given but actual substances, for example martensitic transformations, will be treated in a later paper.

There is 1 Soviet-bloc reference.

ASSOCIATION: Institut fiziki metallov, AN SSSR

(Institute of the Physics of Metals, AS USSR) SUBMITTED: June 16, 1961

Card 2/2

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APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

ARKHAROV, V.I.; VANGENGEYM, S.D.

Effect of internal adserption on the results of precision measurements of polycrystal lattice censtants. Zav.lab/.27 no.6:683-686 '61. (MIRA 14:6)

1. Insitut fiziki metallov AN SSSR i Ural'skiy gesudarstvennyy universitet imeni A.M. Ger'kege, (Adsorption) (Crystal lattices)

S/032/61/027/006/012/018 B124/B203

AUTHORS:

Arkharov, V. I., Sokolova, A. A., and Shangareyev, F. L.

TITLE:

Chamber for collimated X-raying of polycrystalline flat

specimens

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PERIODICAL:

Zavodskaya laboratoriya, v. 27, no. 6, 1961, 751 - 753

TEXT: In the variant of an X-ray chamber (Fig.) developed by the authors, specimen 1 is placed on the object stage 2 and fixed by screw 3. The microsection surface is fixed, and coincides with the vertical front surface of the object stage. The latter can be adjusted vertically and horizontally by two micrometric screws; besides. it can move round the vertical axis together with the frame with the aid of a support in a cylindrical, vertical housing attached to the bench slide. The position with respect to this axis is read from a scale on which zero corresponds to the position of the specimen surface perpendicular to the axis of the collimator 4 which is placed in the casket 5 for taking inverted images. Parallel to the principal optical bench on the base of the chamber,

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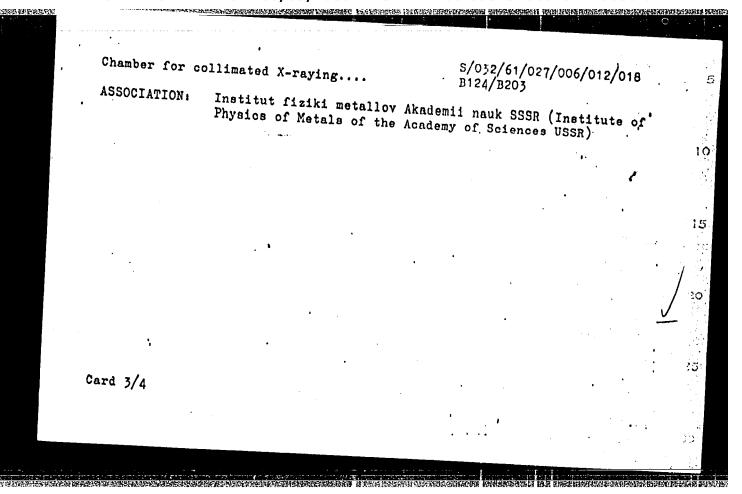
Chamber for collimated X-raying ...

S/032/61/027/006/012/018 B124/B203

there is a second bench 6 over which the slides are moving, which carry the auxiliary stage 7 with two glass prisms 8 with inside total reflection. This stage is moved by the micrometric screw 9. In the back part of the chamber, the microscope 10 is laterally fixed reaching a 15-fold magnification. With the chamber, it is possible to photograph a number of adjacent microsections successively. With the chamber, it is also possible to photograph sufficiently thin specimens in transmitted rays, as well as ground sections with rotation of its surface under an angle to the axis of the primary beam of rays, the angle of swing being read from the rear scale. For this pupose, the semicylindrical casket 11 is screwed to the front side of the frame; the film is placed in this casket. The film is pressed against the casket by the fixing screw 12. Specimens larger than the diameter of the object stage are glued onto the front side of the frame. In collimated X-raying with nicrobeams of rays, a third bench is placed on the bottom of the chamber, parallel to the chamber axis. To this bench, the slide can be fixed which carries the stand for the collimator of the microbeam of rays. There is 1 figure.

Card 2/4

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"



THE PROPERTY OF THE PROPERTY O

S/180/62/000/004/006/009 E040/E435

AUTHORS: Arkharov, V.I., Kuznetsov, E.N. (Sverdlovsk)

TITLE: On the fine structure of crystalline lattice during

polymorphic transformations in cobalt and titanium

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye

tekhnicheskikh nauk. Metallurgiya i toplivo,

no.4, 1962, 143-152

TEXT: Directional bonds are known to exist between crystal lattices of the original and new phases formed during polymorphic transformations in cobalt and titanium; the microstructure of the resultant products of the transformation and their kinetics and temperature dependence differ from that of martensitic transformation in carbon steel, therefore calculations were made of the dimensions and shape of coherent zones present during polymorphic transformations in the crystal lattices of cobalt and titanium. These studies together with available knowledge of the martensitic transformation in steel contribute to the elucidation of the mechanism of the following system of crystal lattices:

Card 1/3

On the fine structure ...

1

S/180/62/000/004/006/009 E040/E435

Face-centred cubic II

Body-centred Close-packed hexagonal cubic

Results are given of calculation of coherent regions for type II and type III transformations in Co and Ti, using essentially the method given previously (Fiz. metallov i metalloved., v.12, no.6, 1961, 853). Basic parameters are tabulated of coherent regions in the β -transition in cobalt (at $450\,^{\circ}\text{C}$) and Ti (at $882.5\,^{\circ}\text{C}$) and the dimensions are given of the β and α -crystal lattices for both metals. Graphical studies were made by superposition of the (111) β and (001) α -phases. It is concluded that the coherency region in Co during transition from the β to α state is in the form of a thin and comparatively narrow but long strip. In Ti, the coherent region is narrower and shorter than in Co, its thickness being approximately of the same order. Comparison of the Card 2/3

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On the fine structure ...

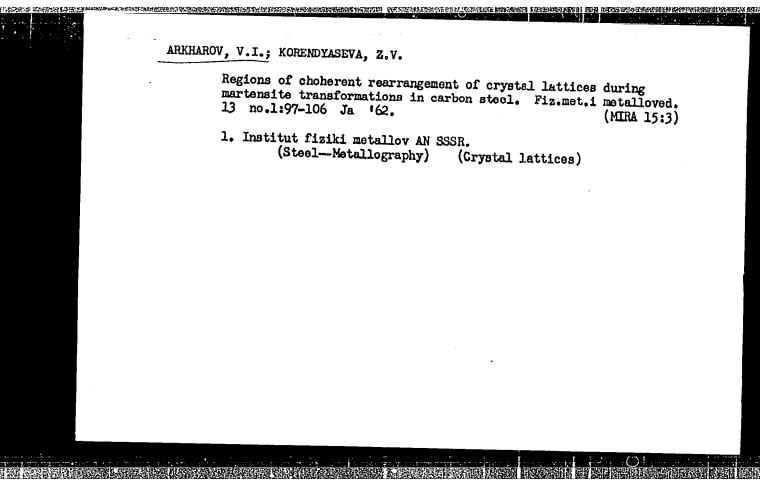
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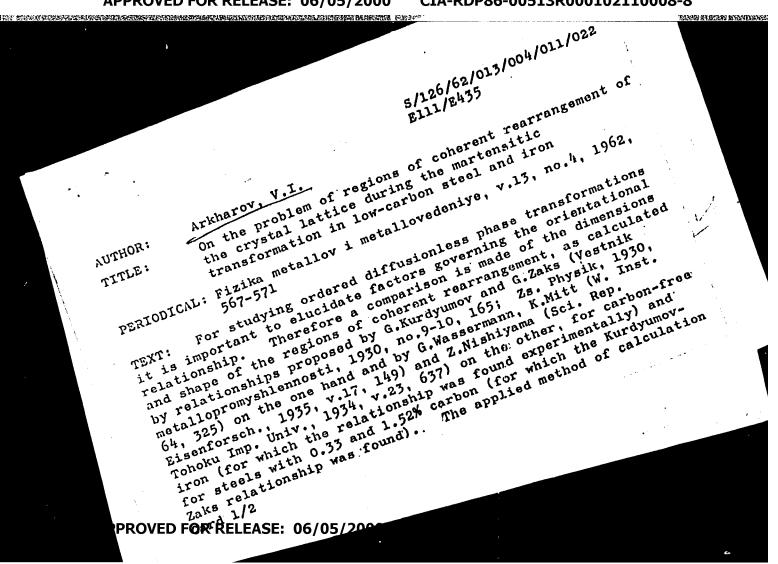
transformations in Co and Ti with the martensitic transformation in steel indicates that the three types of transformation produce essentially different shapes and characters of the coherent regions. The absolute size of the regions is also different. The above differences might provide a basis for studying the causes of the differences in the microstructure of the transformation products and their properties, which depend on the microstructure. There are 6 figures and 5 tables.

SUBMITTED: March 2, 1962

Card 3/3

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On the problem of regions ...

S/126/62/013/004/011/022 E111/E435

is described in a previous paper (FMM, 1961, v.12, no.6). The results show that when the austenite lattice rearrangement follows the Wasserman-Nishiyama mechanism the coherent region is "needle-like". The same happens with the Kurdyumov-Zaks mechanism, but then the needle-length is considerably greater (10 times for austenite with 1.52% C, 5 to 5.8 times for low-Transverse dimensions depend less on the carbon content and are 1.5 to 2 times greater for the Wassermann-Nishiyama mechanism. The needle-axes direction is different in the two mechanisms, as are the axial ratios of the coherent regions (more equiaxial in the Wassermann-Nishiyama). other differences could have a great effect on the formation of the martensite microstructure which proceeds by linkage of adjacent coherent regions. There are 3 figures and 5 tables.

ASSOCIATION: Institut fiziki metallov AN SSSR

(Institute of Physics of Metals AS USSR)

SUBMITTED: November 17, 1961

Card 2/2

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5/126/62/014/001/006/018

E071/E135

AUTHORS:

Arkharov, V.I., Klotsman, S.M., Timofeyev, A.N., and

Trakhtenberg, I.Sh.

TITLE:

An investigation of diffusion properties of

monochalcogenides of transitionary metals. II.

Self-diffusion in polycrystals

PERIODICAL: Fizika metallov i metallovedeniye, v.14, no.1, 1962,

68-74

Since no results of investigations of the laws of TEXT: intercrystalline diffusion in chemical compounds have been published and study of these laws on polycrystalline chemical compounds and their comparison with the laws for elementary substances would give a basis for modelling the structure of intercrystalline linkages in chemical compounds, self-diffusion of Ni in mono- and polycrystals of nickel monosulphide in the temperature range 400-800 °C was studied radiometrically and by autoradiography. Both compounds were obtained by reaction between the individual components in evacuated and sealed ampules, subsequent melting and homogenation. Chemical and X-ray Card 1/2

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An investigation of diffusion ...

S/126/62/014/001/006/018 E071/E135

diffraction analyses confirmed that the specimens were single phased with a structure of the NiAs type. The diffusion was measured on specimens 3-4 mm in diameter and 10 mm thick, one face of which was covered with the diffusion source by vacuo spraying, using Ni53, Co⁶⁰ and Te^{125m} as diffusing elements. Unlike pure metals, predominant self-diffusion in polycrystals along the grain and mosaic block boundaries occurs at temperatures considerably above 0.6-0.7 of the melting temperature and the ratio of intercrystalline diffusion permeability to the "volume" coefficient of self-diffusion amounts to $10^{-1} - 10^{-2}$ cm³/sec. In single crystals of nickel monosulphide predominant diffusion along the grain and mosaic block boundaries occurs at even 0.6 times the melting temperature. The mechanism of scale formation during the process of reaction diffusion can be best studied by measuring the parameters of volume and boundary diffusion of phases entering the composition of the scale. There are 4 figures. ASSOCIATION: Institut fiziki metallov AN SSSR (Institute of Physics of Metals, AS USSR) Card 2/2 SUBMITTED: November 10, 1961.

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APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

JA

ARKHA'	ROV, V.I.
	Submicrostructure and habitus of martensite. Fiz.met.i metalloved. 14 no.5:701-714 N '62. (MIRA 15:12)
	1. Institut fiziki metallov AN SSSR. (Steel-Metallography) (Martensite)

S/136/63/000/002/006/006 E195/E383

AUTHORS: Arkharov, V.I., Borisov, B.S. and Ibragimova, D.M.

TITLE: Gaseous corrosion and embrittlement of technical-grade

nickel

PERIODICAL: Tsvetnyye metally, no. 2, 1963, 72 - 76

TEXT: Tubes made from technical-grade nickel by a process entailing frequent heating of the metal in a gas-filled furnace often show a tendency to cracking. The object of the present investigation was to establish the cause of this fault. The experimental work comprised the following: metallographic examination of specimens of technical-grade nickel and high-purity nickel with small additions of silicon, magnesium, iron, zinc or copper, heated to 1150 - 1200 °C in a gas-filled furnace or in an argon/SO, mixture; X-ray diffraction analysis of nonmetallic phase in an internally oxidized layer formed underneath the oxide scale; impact tests; study of the process of internal oxidation with the aid of a hot-stage microscope. The results can be summarized as follows. 1) The main cause of brittleness of technical-grade nickel tubes is internal oxidation of magnesium and silicon Card 1/2

Gaseous corrosion

S/136/63/000/002/006/006 E193/E383

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introduced initially into the metal during smelting as deoxidizing agents, the embrittling effect of magnesium being more pronounced. 2) No internal oxidation was observed in specimens prepared from pure nickel with up to 0.5% additions of iron, copper or zinc. 3) In the absence of sulphur, the rate of inter- and intragranular internal oxidation was the same. In the presence of sulphur, a low-melting Ni-S eutectic, formed at the grain boundaries, seemed to facilitate the grain-boundary diffusion of oxygen which aggravated the embrittling effect of internal oxidation to such an extent that individual grains broke off the surface layer. 4) The following measures should eliminate or lessen the risk of embrittlement of nickel: replacing magnesium and silicon by other deoxidizing agents such as zinc; preheating the metal in vacuum or in a neutral atmosphere; ensuring that neither the metal nor the furnace atmosphere are contaminated with sulphur. There are 6 figures.

Card 2/2

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

ARKHAROV V. I

AID Nr. 974-14 22 May

DIFFUSION COATINGS AND REFRACTORY COATINGS (USSR)

Metallovedeniye i termicheskaya obrabotka metallov, no. 3, Mar 1963, 59-61. S/129/63/000/003/009/008

The Second Seminar on Diffusion Coatings of Metals and Refractory Compound Coatings on Metallic and Nonmetallic Materials was held 10-12 October 1962 in Odessa. B. N. Arzamasov (Moscow Higher Technical School imeni Bauman) reported on Si-coatings on Mo, W, and Ta and Al-coatings on Mo and W produced by a circulation method in which Si and Al chlorides and gaseous HCl are circulated, i.e., reused. This method improves the quality of coatings and the efficiency and economy of the process. G. N. Dubinin (Moscow Aviation Institute) spoke on the electrical and magnetic properties of "10" and "410" steels diffusion coated with Cr. Mo, W, Si, and T, and of copper and aluminum diffusion coated with Ti, Cr. Si, and B. A. P. Epik discussed C and B diffusion coatings on Ti, Zr, Nb, and Ta, the kinetics of formation of carbide

Card 1/2

AID Nr. 974-14 22 May

DIFFUSION COATINGS [Cont'd]

8/129/63/000/003/009/009

and boride layers, the structure and phase composition of these layers, and some properties for different conditions of coating. V. I. Arkharov speke on the crystallography of phase transformations and reactions. N. V. Titov (Odessa Marine Academy) suggested a method for the approximate calculation of the deformation of a metal during its impregnation with other metals. N. F. Lashko discussed Si diffusion coatings on Nb and Nb-alloys. M. I. Simonova reported on cation distribution in oxides with spinel structure, which are formed on some alloy steels. These data are of great importance for the investigation of diffusion in oxide systems.

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ACCESSION NR: AT4013959

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\$/2659/63/010/000/0239/0246

AUTHOR: Arkharov, V. I.; Konev, V. N.; Nesterov, A. F.; Andrianovskiy, B. P.; Glazkova, I. P.

TITLE: Investigation of metal oxidation in sulfur-saturated air

SOURCE: AN SSSR. Institut metallurgii. Issledovaniya po zharoprochnym splavam, v. 10, 1963, 239-246

TOPIC TAGS: oxidation, sulfur, titanium chromium, maganese, cobalt iron, nickel, metal oxidation, transition element

ABSTRACT: The presence of sulfur in the air frequently leads to acceleration of the oxidation rate, and sometimes to dangerous accidents. The present paper describes the results of investigating the oxidation of Ti, Cr, Mn, Co, Fe and Ni in air containing two chemically active components: oxygen and sulfur. For this group of metals the importance of sulfur in oxidation increases from titanium to nickel. This is explained by the fact that the sulfur activity rises and the oxygen activity drops. The percentage of sulfur in the oxidation scale increases from 0.004% for titanium to complete sulfuration of all the nickel under the layer of NiO. This explains the brittleness of nickel during heat treatment in sulfur-containing media. The process of metal oxidation in sulfur-oxygen media corrobo-

ACCESSION NR: AT4013959

rates the previously published opinions of the authors. Orig. art. has: 2 figures

and 2 tables.

ASSOCIATION: Institut metallurgii AN SSSR (Institute of Metallurgy AN SSSR)

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ARKHAROV, V.I.

Remarks on A.S.Zav'ialov and B.I.Bruk's article "Redistribution of components in solid solutions prior to decay." Fiz.met.i metalloved. 15 no.4:626-628 Ap '63. (MIRA 16:6)

1. Institut fiziki metallov AN SSSR.
(Solutions, Solid) (Crystal lattices)

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

ARKHAROV, V.I.; KUZNETSOV, E.N.

TO THE PARTY OF TH

Shape and size of the regions of coherent rearrangement of crystal lattices during ordered allotropic transformations. Fiz. met. i metalloved. 15 no.5:786-788 My *63. (MIRA 16:8)

1. Institut fiziki metallov AN SSSR.
(Crystal lattices) (Allotropy)

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000102110008-8"

Artivating martensite transformations by preliminary sub-zero treatment of steel. Fiz. met. i metalloved. 15 no.6:929-932 Je *63. (MIRA 16:7) 1. Institut fiziki metallov AN SSSR. (Steel--Netallography) (Phase rule and equilibrium) (metals, Effect of temperature on)

AGAPOVA, Ye.V.; ARKHAROV, V.I., POLIKARPOVA, I.P.

Simultaneous effect of beryllium and antimony impurities on the aging of copper-silver alloys. Fiz. met. i metalloved. 16 no.6:927-929 D '63.

1. Institut fiziki metallov AN SSSR.

(MIRA 17:2)

ARKHAROV, V.I.; BELENKOVA, M.M.; MIKHEYEVA, M.N.; MOISEYEV, A.I.; POLIKARPOVA, I.P.

Change of magnetic susceptibility and the behavior of small impurities in the decomposition of an Ag - Cu solid solution. Izv. AN SSSR. Ser. fiz. 28 no.1:148.151 Ja '64. (MIRA 17:1)

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1. Institut fiziki metallov AN SSSR.

